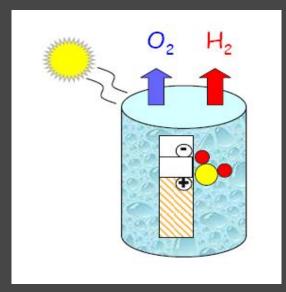
Topic:

Investigation of nanocrystalline diamond films for artificial photosynthesis

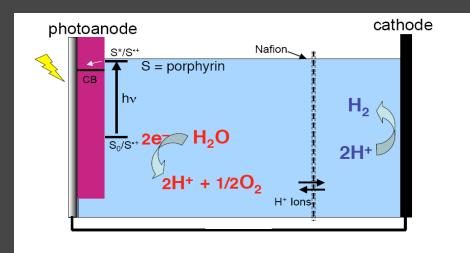
Patras, Greece

Violeta Popova, Christo Petkov



Artificial photosynthesis :

- Replicates the natural process of photosynthesis
- Fuel from sunlight and water (H_2 , CH_4 , CH_3OH)
- Phthalocyanine (catching the sunlight)



Diamond – general properties:

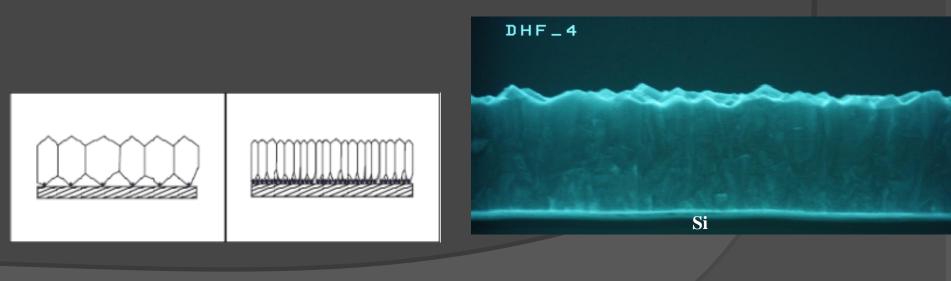


- ✓ the hardest known material
- ✓ chemical inert
- ✓ biocompatible
- ✓ large electrochemical window
- ✓ low coefficient of rubbing
- ✓ high thermal conductivity
- ✓ transparent

\rightarrow Material with unique properties

Nanocrystalline Diamond Films

- polycrystalline diamond films, deposited on substrate
- substrate : diamond (homoepitaxial)
 - other material, e.g.Silicium (heteroepitaxial)
- they have many of the diamond's properties

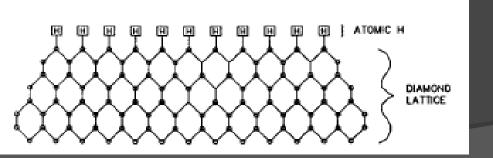


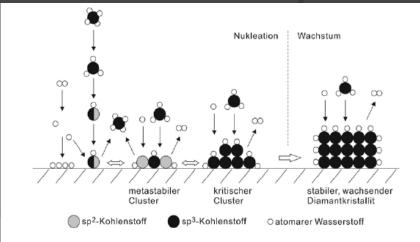
Subjects of the project:

- 1. Investigation of nanocrystalline diamond films as a platform for effective conversion of solar energy
- 2. Investigation and optimization of the properties of nanocrystaline diamond films, like size of the crystalls, morphology, topography and others
- 3. Receiving new knowledges about the interaction between diamond surface and organic molecules (e.g. Porphyrinen)

Chemical Vapour Deposition (CVD)

- activation of plasma \rightarrow Microwave Plasma CVD
- thermal activation \rightarrow Hot Filament CVD
- carbon source \rightarrow ethylen, acetylen, methane and others
- atomic hydrogen
 - selective etching of graphite
 - saturation of the surface



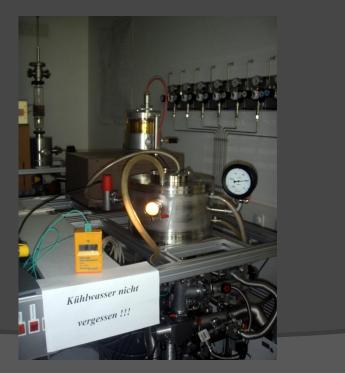


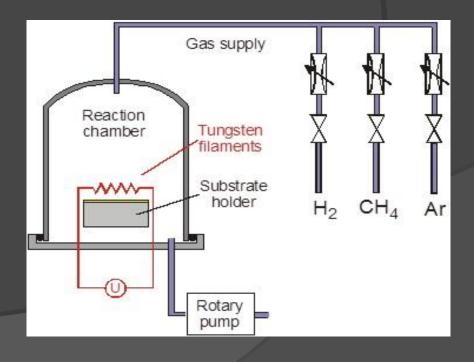
Hot Filament CVD

- filaments with temperature above 2 000°C \rightarrow dissociation of carbon and obtaining of atomic hydrogen

• substrate heating \rightarrow using the filaments

with additional resistive heating



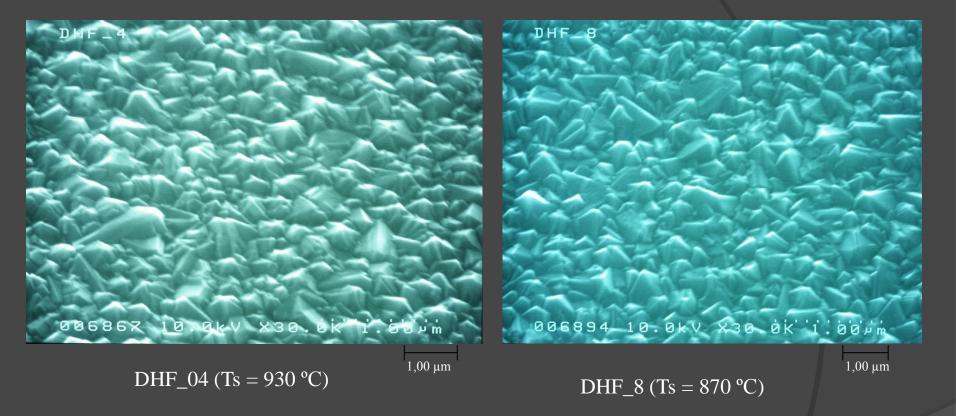


- 1. Changing the technological parameters
- substrate temperature
- nucleation density
- deposition time
- 2. Determine how the morphology, topography and the phases of the diamond films change by:
- low substrate temperature
- low nucleation density
- 3. Apparatuses for investigation:
 - Scanning Electron Microscopy (SEM)
 - Atom Force Microscopy (AFM)
 - Raman Spectroscopy

Deposition parameters

- CH4 / H2 : 1 / 100
- Current Ifl (A) : from 50 to 70 (filaments)
- Tension U(V): 0 120 (substrate heating)
- Pressure : 25 mbar
- Substrate temperature : from 503 °C to 930 °C
- Deposition time : 15 or 180 min
- Pre-treatment : standard or with Slurry-solution

Scanning Electron Microscopy (SEM)



identical surfaces

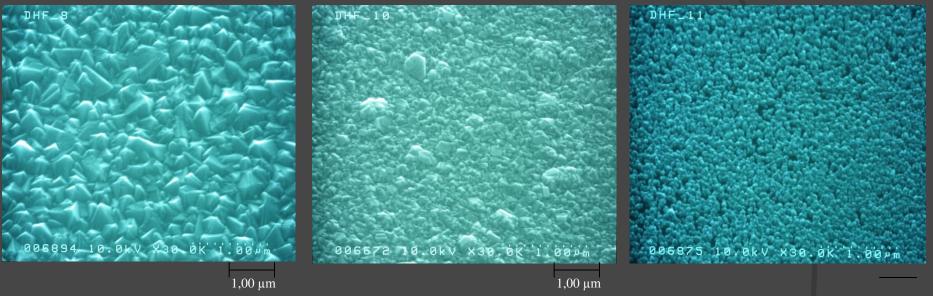
the diamond crystals has the form of a pyramid

SEM

DHF_08 (Ts = 870 °C)

DHF_10 (Ts = 607 °C)

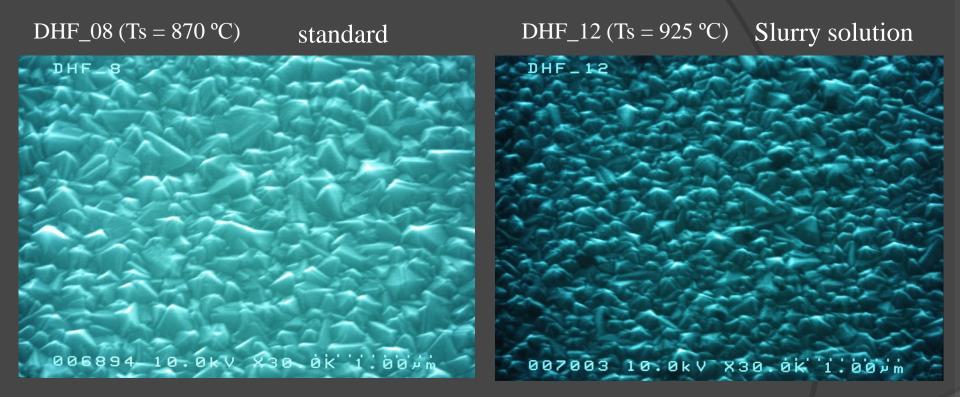
DHF_11 (Ts = 503 °C)



the crystals changed

 transition from nano- to ultrananodiamonds 1,00 µm

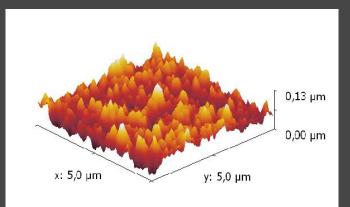
SEM

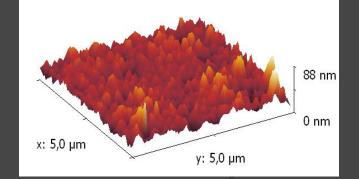


- the crystals are still facetted
- bigger crystals

the nucleation density is lower

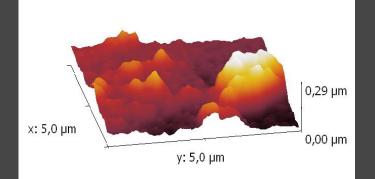
Atom Force Microscopy (AFM)





DHF_01 (Ts = $840 \, ^{\circ}C$)

- the film is closed
- crystals in form of pyramids



DHF_10 (Ts = 607 °C)

- areas with no diamond crystals
- some big structures

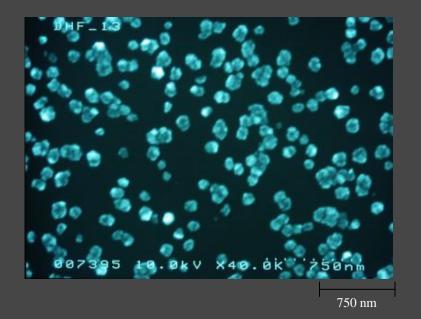
 the crystals are smaller

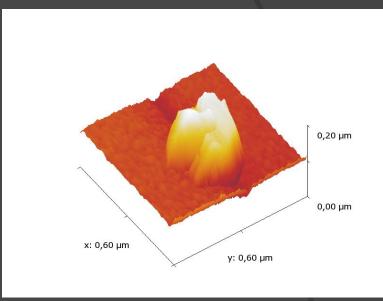
DHF_11 (Ts = $503 \,^{\circ}C$)

 transition from nano- to ultrananodiamonds

SEM and AFM

DHF_13 (Ts = 950 °C), deposition time 15 min \rightarrow single crystals

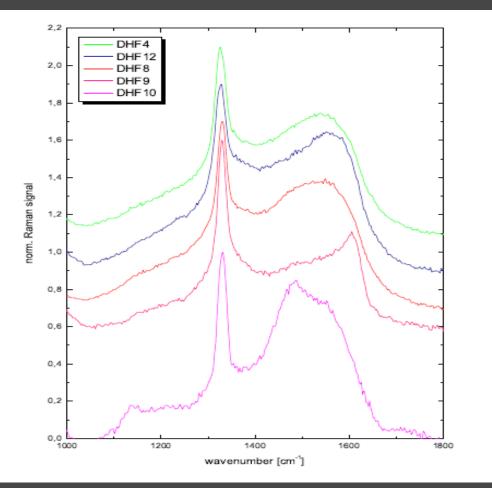




low nucleation density

 the film is not closed → single crystals (15min)

Raman Spectroscopy



- Diamand peak by 1332 cm-1
- G-Peak by 1570 cm-1
 - identical substrate temperatures → identical spectra

930 °C

925 °C

870 °C

705 °C

607 °C

- Peaks by 1150 cm-1 and 1480 cm-1
- Formation of *trans*-Polyacetylen
 - Transition from Nano- to Ultrananodiamonds

Summary:

 preparation and investigation of the properties of nanocrystaline diamond films by:

- different substrate temperature
- different pre-treatment of substrate
- different deposition time
- by low substrate temperature \rightarrow smaller crystals, not facetted
- with standard pre-treatment \rightarrow lower nucleation density
- by low deposition time \rightarrow single crystals

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Thank you for your attention !